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# Formylglycinamide Ribonucleotide Synthetase from *Escherichia coli*: Cloning, Sequencing, Overproduction, Isolation, and Characterization<sup>†,‡</sup>

F. J. Schendel, E. Mueller, and J. Stubbe\*

Department of Chemistry, Massachusetts Institute of Technology, Cambridge, Massachusetts 02139

A. Shiau and J. M. Smith

Seattle Biomed Research Institute, Seattle, Washington 98109-1651 Received July 12, 1988; Revised Manuscript Received November 22, 1988

ABSTRACT: The purL gene of Escherichia coli encoding the enzyme formylglycinamidine ribonucleotide (FGAM) synthetase which catalyzes the conversion of formylglycinamide ribonucleotide (FGAR), glutamine, and MgATP to FGAM, glutamate, ADP, and Pi has been cloned and sequenced. The mature protein, as deduced by the structural gene sequence, contains 1628 amino acids and has a calculated  $M_r$  of 141418. Comparison of the purL control region to other pur loci control regions reveals a common region of dyad symmetry which may be the binding site for the "putative" repressor protein. Construction of an overproducing strain permitted purification of the protein to homogeneity. N-Terminal sequence analysis and comparison of glutamine binding domain sequences (Ebbole & Zalkin, 1987) confirm the amino acid sequence deduced from the gene sequence. The purified protein exhibits glutaminase activity of 0.02% the normal turnover, and NH<sub>3</sub> can replace glutamine as a nitrogen donor with a  $K_{\rm m}=1$  M and a turnover of 3 min<sup>-1</sup> (2% glutamine turnover). The enzyme forms an isolable (1:1) complex with glutamine:  $t_{1/2}$  is 22 min at 4 °C. This isolated complex is not chemically competent to complete turnover when FGAR and ATP are added, demonstrating that ammonia and glutamine are not covalently bound as a thiohemiaminal available to complete the chemical conversion to FGAM. Hydroxylamine trapping experiments indicate that glutamine is bound covalently to the enzyme as a thiol ester. Initial velocity and dead-end inhibition kinetic studies on FGAM synthetase are most consistent with a sequential mechanism in which glutamine binds followed by rapid equilibrium binding of MgATP and then FGAR. Incubation of [180] FGAR with enzyme, ATP, and glutamine results in quantitative transfer of the <sup>18</sup>O to P<sub>i</sub>.

Formylglycinamide ribonucleotide (FGAM)<sup>1</sup> synthetase is the fourth step in the purine biosynthetic pathway and cata-

<sup>†</sup>The nucleic acid sequence in this paper has been submitted to Gen-Bank under Accession Number J02848.

lyzes the irreversible conversion of formylglycinamide ribonucleotide (FGAR), glutamine, and ATP to FGAM, P<sub>i</sub>, ADP, and glutamate (eq 1). This protein was first purified to homogeneity from Salmonella typhimurium by French et al.

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<sup>&</sup>lt;sup>1</sup> Abbreviations: FGAM, formylglycinamidine ribonucleotide; FGAR, formylglycinamide ribonucleotide; HEPES, N-(2-hydroxyethyl)-piperazine-N'-2-ethanesulfonic acid; SDS, sodium dodecyl sulfate; AIR, aminoimidazole ribonucleotide.

in 1963 and then from chicken liver by Mizobuchi and Buchanan in 1968. Recently, Ebbole and Zalkin (1987) reported the cloning and characterization of a 12-gene cluster in Bacillus subtilis encoding nine enzymes for the purine biosynthetic pathway including the gene, purL, for FGAM synthetase. The proteins from S. typhimurium and chicken liver are monomers of  $M_r = 135$ K and 133K, respectively. This contrasts with results from the gene sequence of B. subtilis (identified by sequence homology with the E. coli protein reported in this paper) which indicate that FGAM synthetase is composed of two polypeptides:  $M_r = 24755$  (purQ gene) and 80 300 (purL gene).

Extensive studies from Buchanan's laboratory on the chicken liver protein have provided much insight about the nature of the glutamine binding domain (Buchanan, 1973) and have provided some thought-provoking information about the properties of the ATP and FGAR binding site(s). Specifically, Mizobuchi et al. (1968) made the unusual observation that chicken liver FGAM synthetase forms a stable isolable complex with glutamine,  $t_{1/2} = 125$  min at 4 °C, and a stable isolable complex with FGAR and ATP,  $t_{1/2} = 62$  min at 4 °C.

Recent efforts have focused on unraveling a generalized mechanism by which enzymes in the purine [aminoimidazole ribonucleotide (AIR) synthetase] and pyrimidine (CTP synthetase) pathways convert amides to amidines using ATP as a dehydrating agent (Schendel & Stubbe, 1986; Schrimsher et al., 1986; von der Saal et al., 1985; Westheimer, 1981) and by which enzymes from these pathways convert glutamine to a "putative NH<sub>3</sub> equivalent" (Zalkin, 1985).

The unusual stability of the FGAM synthetase-substrate complexes reported for the chicken liver enzyme might provide a opportunity to examine the detailed mechanism if these complexes are chemically and kinetically competent in the presence of the additional required substrate(s).

Initially we focused our efforts on obtaining large amounts of the chicken liver protein for use in detailed mechanistic studies. Unfortunately, instability of the protein as well as lack of reproducibility of isolation of the protein, in analogy with problems reported by Mizobuchi and Buchanan (1968) and French et al. (1963), made an alternative source of the protein desirable.

The availability of Escherichia coli purL mutants has provided us with this alternative source. The present paper reports the successful cloning behind the  $\lambda$ pL promoter and sequencing of the gene for FGAM synthetase and overexpression of the protein. FGAM synthetase subsequent to heat induction is 50% of the total protein (approximately 150 times that found in wild-type  $E.\ coli$ ), which has greatly facilitated rapid isolation or large amounts of homogeneous protein. The kinetic mechanism of this protein has been elucidated by initial velocity studies in conjunction with product and dead-end inhibition studies. A detailed study of the properties of the FGAM synthetase–glutamine complex is also reported. These studies provide the foundation for future detailed investigations

on the regulation of gene expression and on the mechanism of conversion of FGAR to FGAM.

### MATERIALS AND METHODS

Sephadex G-25, Sephadex G-50, Sephadex G-150, Sephacryl S-200, DEAE-Sephadex A-25 and A-50, phosphoenolpyruvic acid, ATP, ampillicin, L-glutamine, rabbit muscle pyruvate kinase (355 units mg<sup>-1</sup>), rabbit muscle L-lactate dehydrogenase (920 units mg<sup>-1</sup>), E. coli glutaminase (8.4 units mg<sup>-1</sup>), bovine serum albumin, and molecular weight standards for gel filtration and gel electrophoresis were obtained from Sigma Chemical Co. Nitrocellulose, DEAE-agarose Bio-Gel A, hydroxylapatite HPT, goat anti-rabbit IgG-horseradish peroxidase conjugate, Tween-20, and horseradish peroxidase color reagent were purchased from Bio-Rad. Cellulose F thin-layer chromatography plates were obtained from J. T. Baker Chemical Co. [1-13C]Glycine (90.0 atom % 13C) was obtained from MSD Isotopes. H<sub>2</sub><sup>18</sup>O (95.1 atom % <sup>18</sup>O) was purchased from Monsanto Research Corp. [U-14C]-L-Glutamine (283 mCi/mmol) was purchased from New England Nuclear. [1-13C, 18O2]Glycine was prepared by the procedure of Mears and Sobotka (1939). FGAR was synthesized by the procedure of Schendel and Stubbe (1986). Concentrations of protein solutions were determined by the method of Lowry et al. (1951) using bovine serum albumin as a standard. L-Glutamine concentrations were determined by measuring the ammonia released or the glutamate formed by the action of glutaminase (Williamson & Corkey, 1979). AIR synthetase (3 units mg<sup>-1</sup>) was isolated from E. coli TX393/pJS24 by the procedure of Schrimsher et al. (1986a). 13C NMR spectra were obtained at 50.3 MHz, and <sup>31</sup>P NMR spectra were obtained at 80.9 MHz on a Nicolet 200-MHz broad-band spectrometer. All solutions were analyzed for radioactivity in Packard Scint-A scintillation fluid with a Packard 310 liquid scintillation counter. Protein sequencing was done according to the automated Edman degradation procedure by the University of Wisconsin Biotechnology Center.

Strains and Media. Strain TX160 [(\Delta lac) purL213::lac-ZY::λp1(209)] was created from strain TX40 (Smith & Gots, 1980) by the procedure of Casadaban (Groisman et al., 1984). Strain TX540 [ $\Delta(lac)$  purL213::lacZY:: $\lambda p1(209)srlC300$ :: Tn10 recA56] was constructed and employed for complementation analysis. Strain JM83 (Messing, 1979) was used as a recipient to identify subclones containing restriction fragments from the purL region while strain JM101 (Messing, 1979) was employed for the propagation of M13 bacteriophages. Strain TX635 (F' lacZ+ cI857; Mieschendahl & Müller-Hill, 1985) contains an episome-borne temperaturesensitive  $\lambda$  repressor and was used as a host for the  $\lambda pL$ plasmids. Strains were made competent and transformed by the procedure of Dagert and Ehrlich (1979). The minimal medium of Neidhart et al. (1974) and the rich media described by Miller (1972) were used for the growth of the E. coli K12 strains. The recombinant DNA techniques employed have been described (Tiedeman et al., 1985).

Purification of E. coli FGAM Synthetase. E. coli strain TX635/pJS113 was grown at 30 °C in LB media (Mantiatis, 1982) in house distilled water supplemented with 1 mL/L Vogel-Bonner (1956) trace minerals. Bacteria were grown to an OD<sub>600</sub> = 0.95 (doubling time, 1 h), heat induced by an equal volume of LB media at 56 °C, and grown at 42 °C for an additional 3 h. The bacteria were harvested, and the cells were frozen with liquid nitrogen and stored at -80 °C. The purification of FGAM synthetase was carried out at 4 °C, and all buffers used contained 1 mM glutamine and 5 mM  $\beta$ -mercaptoethanol.

The cells (1 g) were resuspended in 5 volumes (5 mL) of 50 mM potassium phosphate, pH 6.8, and broken in a French press pressure cell at 12000 psi, and the extract was centrifuged at 17000g for 20 min. The supernatant was adjusted to 40% saturation in ammonium sulfate by the slow addition of solid ammonium sulfate (0.243 g/mL). After the ammonium sulfate had dissolved, the solution was stirred for 20 min and then centrifuged at 17000g for 10 min. The supernatant was then brought to 55% saturation of ammonium sulfate by the addition of solid ammonium sulfate (0.097 g/mL). The solution was again stirred for 20 min and then centrifuged at 17000g for 10 min. The pellet was redissolved in a minimal volume (0.8 mL) of 50 mM potassium phosphate, pH 6.8, and then desalted on a Sephadex G-25 column (2 × 15 cm) equilibrated in the same buffer. The fractions containing protein were collected and applied to a column of DEAEagarose (1.5 × 8.5 cm) equilibrated in 50 mM potassium phosphate, pH 6.8. The column was washed until the absorbance at 280 nm was less than 0.01. The FGAM synthetase was then eluted with 50 mM potassium phosphate (pH 6.8),  $75 \times 75$  mL linear gradient from 0 to 150 mM KCl. The fractions containing activity were pooled and concentrated to less than 5 mL in an Amicon ultrafiltration apparatus using a PM-30 membrane. The enzyme, quick frozen in 50 mM potassium phosphate, pH 6.8, containing 25% (v/v) glycerol, was stored at -80 °C and was stable for at least 2 weeks.

Native and Subunit Molecular Weight of FGAM Synthetase. A column of Sephadex G-150 was equilibrated with 50 mM potassium phosphate, pH 6.8, and calibrated with blue dextran (average  $M_r$  of  $2 \times 10^6$ ) and proteins of known molecular weight. The proteins used were yeast alcohol dehydrogenase ( $M_r = 150000$ ), conalbumin ( $M_r = 78000$ ), ovalbumin ( $M_r = 45\,000$ ), bovine erythrocyte carbonic anhydrase ( $M_r = 29\,000$ ), and horse heart cytochrome c ( $M_r =$ 12 400). The native  $M_r$  of FGAM synthetase was obtained from a plot of  $M_r$  vs  $K_{ave}$ . Subunit molecular weight was determined from SDS-polyacrylamide gel electrophoresis (7.5% acrylamide) performed according to the procedure Laemmli (1970). The molecular weight standards used included E. coli  $\beta$ -galactosidase ( $M_r = 116000$ ), rabbit muscle phosphorylase B ( $M_r = 91100$ ), bovine serum albumin ( $M_r$ = 60 000), egg albumin ( $M_r$  = 44 000), and bovine erythrocyte carbonic anhydrase ( $M_r = 29000$ ).

Enzyme Assays. FGAM synthetase was assayed by one of three methods. (1) The amount of FGAM produced was determined according to a coupled assay procedure with AIR synthetase (Schendel & Stubbe, 1986) in which the buffer was 50 mM HEPES (pH 7.25). (2) The formation of ADP was quantitated spectrophotometrically by a coupled assay procedure with pyruvate kinase and lactate dehydrogenase. A typical reaction mixture contained in a final volume of 1 mL 40 mM HEPES (pH 7.25), 20 mM MgCl<sub>2</sub>, 10 mM KCl, 1 mM L-glutamine, 2 mM ATP, 0.2 mM  $\alpha/\beta$ -FGAR, 0.2 mM NADH, 1.0 mM phosphoenolpyruvate, 3.5 units of pyruvate kinase, and 4.7 units of lactate dehydrogenase, preincubated at 37 °C. The reaction was started by the addition of FGAM synthetase (0.02 unit). The initial velocities were calculated from the decrease in absorbance at 340 nm with  $\epsilon = 6200 \text{ M}^{-1}$ cm<sup>-1</sup>. (3) The formation of glutamate was determined according to a coupled assay procedure with glutamate dehydrogenase in a stepwise manner (Schendel & Stubbe, 1986). One unit of enzymatic activity is defined as the amount of enzyme required to produce 1  $\mu$ mol of product min<sup>-1</sup> at 37 °C.

Kinetic Analysis. All kinetic studies were conducted in the presence of 40 mM HEPES (pH 7.25), 20 mM MgCl<sub>2</sub>, and

10 mM KCl. The Michaelis constants for FGAR, MgATP, and glutamine were determined by varying the concentration of one substrate at saturated levels of the other two and fitting the data with nonlinear least-squares regression analysis to eq 2. Initial velocity data were obtained by varying the con-

$$v = VA/[K_a + A] \tag{2}$$

centration of two substrates from  $0.2K_{\rm m}$  to  $5K_{\rm m}$  while holding the third substrate concentration constant  $(1.5K_{\rm m})$  or by varying the concentration of one substrate at a fixed ratio of the other two (Fromm, 1967; Chu & Henderson, 1972). The intial velocity data were fit to eq 3 or the corresponding

$$v = VABC/[\text{const} + (\text{coeff A})A + (\text{coeff B})B + (\text{coeff C})C + K_aBC + K_bAC + K_cAB + ABC]$$
(3)

equation with the appropriate terms missing, to determine the best steady-state kinetic model (Viola & Cleland, 1982). Inhibition studies were analyzed with the computer programs of Cleland (1979). Data corresponding to competitive, noncompetitive, or uncompetitive inhibition were fit to eq 4-6.

$$v = VA/[K_{a}(1 + I/K_{is}) + A]$$
 (4)

$$v = VA/[K_a(1 + I/K_{is}) + A(1 + I/K_{ii})]$$
 (5)

$$v = VA/[K_a + A(1 + I/K_{ii})]$$
 (6)

Exchange Reactions. The ATP/ADP exchange reactions contained 40 mM HEPES (pH 7.2), 20 mM MgCl<sub>2</sub>, 10 mM KCl, 1.4 mM MgATP, 1.2  $\mu$ M [U-14C]ADP (1.0 × 106 cpm/µmol), and 0.02 unit of FGAM synthetase in a final volume of 200  $\mu$ L. The reaction mixture was incubated at 37 °C, and  $50-\mu$ L aliquots were withdrawn at 0, 5, 10, and 20 min. These samples were added to 100 µL of 50 mM potassium phosphate, pH 4.8, containing 5 mM tetrabutylammonium bromide and immediately frozen in a dry iceacetone bath. The samples were analyzed by reverse-phase ion-pairing HPLC on an Alltech C-18 column with a flow rate of 1 mL/min using a solvent composed of 50 mM potassium phosphate, pH 4.8, 5 mM tetrabutylammonium bromide, and 24% (v/v) methanol. The retention time for ADP was 8.5 min and for ATP was 16 min. The ATP peak was collected and concentrated to dryness in vacuo, redissolved in 1 mL of water, and analyzed for radioactivity by liquid scintillation counting.

The glutamine/glutamate exchange reactions contained 40 mM HEPES (pH 7.2), 20 mM MgCl<sub>2</sub>, 10 mM KCl, 460  $\mu$ M glutamine, 482  $\mu$ M [U-<sup>14</sup>C]glutamate (1.7 × 10<sup>6</sup> cpm/ $\mu$ mol), and 0.02 unit of FGAM synthetase in a final volume of 200  $\mu$ L. The reactions were incubated at 37 °C, and 40- $\mu$ L aliquots were withdrawn at 0, 5, 10, and 20 min. The samples were applied to a 1-mL DEAE-Sephadex A-25 column. The column was washed with 5 mL of water to elute the glutamine, and 2 mL of the eluant was analyzed for radioactivity by liquid scintillation counting.

Formation of Enzyme–Glutamine Complex. To a total volume of 300  $\mu$ L containing 40 mM HEPES, pH 7.25, 10 mM KCl, 20 mM MgCl<sub>2</sub>, and 3 mM [U-<sup>14</sup>C]-L-glutamine (3.5 × 10<sup>6</sup> cpm/ $\mu$ mol) was added 1–10 nmol of FGAM synthetase, and the mixture was incubated at 37 °C for 5 min. The reaction was then applied to a Sephadex G-50 column (0.75 × 21 cm) equilibrated in 50 mM HEPES, pH 7.3. Fractions of 1 mL were collected, the absorbance at 280 nm was recorded, and 100–500  $\mu$ L of each fraction was analyzed for <sup>14</sup>C by liquid scintillation counting. Protein did not cause any significant quenching during scintillation counting. The molar ratio of [U-<sup>14</sup>C]glutamine to enzyme (binding ratio) was calculated with a  $M_r$  = 140K for the protein, based on the gene sequence, and 2 units mg<sup>-1</sup> as the specific activity of the pure

protein. The half-life of the enzyme-glutamine complex was determined by using columns of varying lengths  $(0.75 \times 15-45 \text{ cm})$  run with variable flow rates and by plotting the log of the binding ratio vs the time the complex was on the column.

Reaction of Enzyme-Glutamine Complex with Hydroxylamine. To 5.0 or 6.0 nmol of complex (isolated as discussed above) in 500  $\mu$ L of 50 mM HEPES, pH 7.3, was added 500 μL of 2 M hydroxylamine, pH 7.3, or 1.0 mL of 1.5 M hydroxylamine in 6 M guanidine hydrochloride. The reactions were incubated at 25 °C for 20 min and then diluted to 20 mL and applied to DEAE-Sephadex A-25 columns (1  $\times$  5 cm and  $1 \times 10$  cm, respectively). The columns were washed with water, conditions to elute glutamine and  $\gamma$ -glutamylhydroxamate, and then with 250 mM triethylammonium bicarbonate, conditions to elute glutamate. Fractions of 1 mL were collected and analyzed for radioactivity by liquid scintillation counting. Fractions from the H<sub>2</sub>O elution containing radioactivity were pooled, adjusted to pH 7.0, and heated to 100 °C for 7.5 min, a procedure which converts  $\gamma$ -glutamylhydroxamic acid to pyrrolidonecarboxylic acid (Levintow et al., 1955). To separate any resulting pyrrolidonecarboxylic acid from glutamine, the sample was diluted to 10 mL with water and applied to a DEAE-Sephadex A-25 column (1 × 5 cm). The column was washed with 10 mL of water, and eluted with 250 mM triethylammonium bicarbonate. The fractions containing radioactivity were pooled, concentrated to dryness, and utilized for further characterization.

Separation of L-Glutamate, L-Glutamine, and Pyrrolidonecarboxylic Acid. Samples isolated by the preceding protocol were spotted on cellulose F plates and developed with a solvent composed of absolute ethanol, 2-methyl-2-propanol, water, and formic acid (88%) (60:20:15:5) (Mizobuchi & Buchanan, 1968a). R<sub>f</sub> values were as follows: L-glutamine, 0.25; L-glutamate, 0.36; pyrrolidonecarboxylic acid, 0.62. Pyrrolidonecarboxylic acid was visualized according to the starch-iodide procedure of Pan and Dutcher (1956), and amino acids were visualized with 0.3% ninhydrin in acetone. The radioactivity was quantitated by liquid scintillation counting subsequent to cutting the plate into 1-cm strips and elution of compounds with 1 mL of 200 mM triethylammonium bicarbonate.

Turnover of Enzyme–Glutamine Complex. To 6.8 nmol of complex (isolated as described above) in 600  $\mu$ L of 10 mM HEPES, pH 7.3, was added 100  $\mu$ L of 200 mM HEPES, pH 7.3, containing 150 mM MgCl<sub>2</sub>, 75 mM KCl, 15 mM MgATP, 3.5 mM FGAR, and 0.2 unit of AIR synthetase. The reaction mixture was incubated at 37 °C for 5 min, and the amount of AIR produced was determined according to a modified Bratton–Marshall assay ( $A_{500}$  nm and with  $\epsilon$  = 24 600 M<sup>-1</sup> cm<sup>-1</sup> (Schendel & Stubbe, 1986; Schrimsher et al., 1986b). A control reaction containing 4.8 nmol of complex was run under identical conditions with those described above except 2.1 nmol of glutamine was also included. This control allows determination of the lower limit of detection of AIR by this method.

<sup>18</sup>O Transfer from [<sup>18</sup>O]FGAR to  $P_i$ . To a 5-mm NMR tube containing 30 mM ATP, 30 mM L-glutamine, 10 mM KCl, 50 mM MgCl<sub>2</sub>, and 40 mM HEPES, pH 7.3, in a final volume of 1 mL (30% D<sub>2</sub>O) was added 1.0 unit of FGAM synthetase. The <sup>31</sup>P NMR spectrum was taken at 80.9 MHz, and then 15  $\mu$ mol of [<sup>18</sup>O]FGAR was added and the reaction mixture incubated at 37 °C for 30 min. When the reaction was complete, determined by <sup>31</sup>P NMR, the reaction mixture was diluted to 3 mL and passed through an Amicon Centricon-30 ultrafiltration apparatus with a YM-30 membrane.

The filtrate was collected and applied to a DEAE-Sephadex A-25 column (1.0  $\times$  4 cm). The column was washed with 2 volumes (8 mL) of water, and the inorganic phosphate was eluted with 250 mM triethylammonium bicarbonate (pH 7.8). The fractions containing the inorganic phosphate were pooled, concentrated to dryness in vacuo, and redissolved in 0.5 mL of D<sub>2</sub>O containing 1 mM EDTA. The sample was placed in a 5-mm NMR tube, and the <sup>31</sup>P NMR spectrum was recorded at 80.9 MHz with a sweep width of  $\pm$ 500 Hz, quadrature phase detection, a 90° pulse angle with a pulse delay of 2 s, and an 8K data block. To enhance resolution, exponential multiplication with a line-broading factor of 0.1 Hz was applied to the FID before Fourier transformation. The ratio of [<sup>16</sup>O]/[<sup>18</sup>O]P<sub>i</sub> was determined by weighing the appropriate peaks.

Western Blot Analysis. SDS-polyacrylamide gel electrophoresis was run as described above, and the proteins were transferred to nitrocellulose electrophoretically (Burnette, 1981). Antibodies used in probing the blots were prepared in rabbits by injection of 300-500 µg of native chicken liver FGAM synthetase in Freund's complete adjuvant followed by booster injections with 300-500 µg of FGAM synthetase in Freund's incomplete adjuvant. Antibody was visualized by using a goat anti-rabbit IgG-horseradish peroxidase conjugate as described by Hawkes et al. (1982). Proteins were visualized by Coomassie blue staining for 10 s. Antibody 80 times the concentration required to detect the chicken liver enzyme indicated no cross-reactivity.

#### RESULTS AND DISCUSSION

Cloning and Complementation Analysis of the purL Locus. Plasmids containing the purL locus were cloned by the mini-Mu procedure of Casadaban (Groisman et al., 1984) and isolated by complementation of strain TX540. After initial characterization of several plasmids, the purL locus was subcloned by partial Sau3A digestion into the BamHI site of plasmid pUC19 (Yanisch-Perron et al., 1985). A representative subclone with an approximately 6-kb insert was designated pJS80 and used for further analysis. The limits of the purL structural gene were localized by testing specific subcloned restriction fragments for complementation in strain TX540. One end of the purL gene was localized between a MluI and a SspI site while the other end was localized between a FspI and a SspI site (Figure 1). This region can maximally encode a protein of approximately 150 000 daltons, sufficient to accommodate the subunit size of FGAM synthetase from Salmonella (French et al., 1963).

Construction of an Expression Vector for FGAM Synthetase. An expression vector for FGAM synthetase was created by reconstructing a DNA fragment spanning the HindIII site at nucleotide 1273 to the SspI site at nucleotide 5339. This fragment, lacking the purL promoter, was then inserted into plasmid pJS88 (A. A. Tiedeman and J. M. Smith, unpublished results) to produce pJS113. Plasmid pJS88 is a λpL expression vector similar to the ones described by Remaut et al. (1981). Strain TX635 was used as the host, and synthesis of FGAM synthetase was induced by a temperature shift to 42 °C.

DNA Sequence Analysis. DNA sequence was determined by the dideoxy chain termination method of Sanger (Sanger et al., 1977). The 6-kb DNA fragment containing the purL locus was subcloned as various separate restriction fragments into M13mp18 and overlapping deletion subclones generated by the procedure of Henikoff (1984) for DNA sequencing. The DNA sequence of the entire 6-kb DNA fragment was assembled from overlapping subclones for both strands as shown in Figure 2. The DNA sequences were compiled and

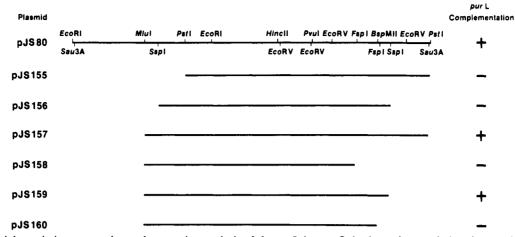


FIGURE 1: Partial restriction map and complementation analysis of the purL locus. Only the major restriction sites pertinent to this study are shown. The Sau3A sites represent the termini of the restriction fragment while the outer EcoRI and PstI restriction sites show the orientation within the cloning region of pUC19. Complementation studies were carried out in strain TX540.

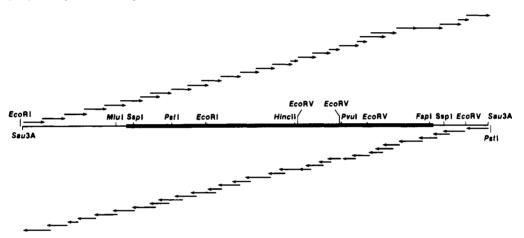


FIGURE 2: Sequencing strategy for purL. The location of the 6-bp restriction enzyme sites used in subcloning and sequencing studies are indicated. The arrows denote sequencing direction, and the length of the arrow is proportional to the number of nucleotides determined. The purL coding region extends from nucleotide 1294 to nucleotide 5178 and is indicated by the thickened line.

analyzed by computer (Larson & Messing, 1983) and by the facilities of the BIONET resource.

Derived Amino Acid Sequence. After the sequence of the 6-kb DNA fragment containing the purL locus was determined, it was analyzed for open reading frames. Within the boundaries indicated by the complementation analysis, only one open reading frame capable of encoding a protein of greater than 100 000 daltons was found. This open reading frame extends from nucleotide 1294 to nucleotide 5178 (Figure 3) and encodes a protein of 141 418 daltons. The start of the purL open reading frame was confirmed by comparison to the amino-terminal sequence of purified FGAM synthetase, obtained by automated Edman degradation of the purified protein:2

This open reading frame is preceded by a good ribosomal initiation site (Shine & Dalgarno, 1974), GAG, at nucleotides 1285-1287 (Figure 3).

An additional confirmation of the correctness of the FGAM synthetase sequence is indicated by the finding of nearly

identical sequences corresponding to the glutamine binding domain of both chicken liver and Salmonella typhimurium FGAM synthetases. This region, nucleotides 4684-4710, is overlined in Figure 3.

Identification of purL Control Region. After identification of the purL coding region, a 20-mer oligonucleotide designed to anneal to nucleotides 1324-1343 within the purL coding region was synthesized. After annealing to total cellular RNA extracted from strain TX540 containing plasmid pJS80, primer extension with reverse transcriptase was carried out as described by Miller et al. (1986). The size of the primer extension product (Figure 4) indicated nucleotide 1248 as the most likely initiation point. Seven nucleotides upstream of the transcription initiation site, a consensus -10 region is found (Harley & Reynolds, 1987), and 15 nucleotides upstream of the -10 region, a good -35 region (four of six matches) is found (Raibaud & Schwartz, 1984).

Comparison of the purL control region to the other pur loci control regions revealed a common region of dyad symmetry (nucleotides 1204-1216) (Smith & Daum, 1986; Tso et al., 1982). This sequence is located sufficiently close to the -35 region such that it may act as a binding site for a regulatory protein (the putative repressor protein) which controls transcription by altering binding of RNA polymerase.

Isolation of E. coli FGAM Synthetase. FGAM synthetase from E. coli strain TX635 containing pJS113 has been purified to homogeneity, Table I. From 1 g of cells, 23 mg of homogeneous protein was isolated in 60% overall yield.

<sup>&</sup>lt;sup>2</sup> Two sequences occurred. The lower-case letters show less confidence. In fact, the first (a) round did contain R, and the (l) round contained serine.

EC\_purL

10 20 30 40 50
GATCCGTART GCCGAGGCTT TGCGCGGTAT TTTTGGTTAR CATCATCATG 60 70 80 90 100
CCGCGCACAC CCGTCGGTGA AGTGGCCTGT GCATCCCAGT GCGATTCCTG 110 120 130 140 150
ATAMGCANTA GCGGCCAGCA ANCGCCAGTC ANTITCTTCG GCGTATTTCT 160 170 180 190 200
CARACAGGGG CTTTARCTGC GGCAGTACCG CATCGACGGC GCGTARRANT 210 220 230 240 250 GTGGGGGTAT CGACGTAATC ARATCATCG CCATGCCCCA GGTATTTCTC 260 270 280 290 300
TTCNATGCGT GCCAGCGTAC CGTCTTCATT CATTTCGTTG AAGAAGTCGA 310 320 330 340 350 GCAGGGCGGC GGAAAGGGTA TTATCGCCAT CTAACGGGCT AAACCAAGTC 360 370 380 390 400 ACCGGTTGTT CATCGGTGAT ATCGAGCGCT ACGGCGAGCT CCGGGTGAAC GCGCTGAAAC AGGCTGATGG CGACAGAATC AGCAATGGTG TAATCGAGTT 460 470 480 490 500 TTCCTTCGAT GACATCTTCC ATTACTCCG CAGAGCCTTT TTTGTCGTCT 530 ACCTTCCAGC TTAATTCCGG GAATTTTGTT TCTTTCAGGG TCTGGAGATC 560 570 580 590 600 GTTAACCACC ACATGACCCG GTGCAACGGT GAGTCTGTCC GCCGTCAGGT TGCCCAGCGT ACGTGGGCGA TACTGACCCA CTTTATAAAC CAGTTGTTGT GACACGGAAT AATAGGTAGG GCCAGGCTGA TAATTTTTTA CCCGCTCACT 720 730 GTTATAGACA AGTCCTGCCG CCAGCAGGTC GGCGTTACCA TTATCAAGGT CGTCAAACAG CTGGCTGATA TTCTGCCGCA CGGTCACTTT CAGTTTTACG CCGAGGTAAT CGGCAAACTG TTTCGCCAGT TCGTAATCCA GGCCAAAAGG 890 870 880 TTTCCCGTTG ATTTCGTTAT AAGTCAGGGG AGTATGAATG GTGCTCACAC 910 920 930 940 950 GCAACTCTCC CCGCGCTTGA ATGGCGGCGA TACGGTTGTC GGCTTTACCA AACCAGGGAA TGGATGGCCA GAGAGCGACC GCGAGCAGCA GTGCCAGAAT 1010 1020 1030 1040 1050 GCCGATGARC AGATRATTAA TCTTTAATTT TTTCAATTAG TTAATTCTCT 1070 1080 GTGTCGTGCG CGTCCCAGCT TGAAAAAACG TAATAATAGT GAAAGGTTTA 1120 CTCATAAATG AGCGGCATTT TGCGTAAACC TGCGCCAGAT GGCAACTTAT TACAGCCATT GGCGGCACGC GTTGCTAATT CACGATGGTG ATTTTATTTC 1220 1230 CACGCAAACG GTTTCGTCAG CGCATCAGAT TCTTTATAAT GACGCCCGTT 1270 1280 TCCCCCCCTT GGGTACACCG ARAGCTTAGA AGACGAGAGA CTT ATG ATG
B/D MET MET GAA ATT CTG CGT GGT TCG CCT GCA CTG TCG GCA TTC CGA ATC
Glu Ile Leu Arg Gly Ser Pro Ala Leu Ser Ala Phe Arg Ile 1353
AAC AAA CTG CTG GCA CGT TTT CAG GCT GCC AGG CTC CCG GTT
Asn Lys Leu Leu Ala Arg Phe Gln Ala Ala Arg Leu Pro Val 1398 1413 CAC AAT ATT TAC GCC GAG TAT GTC CAT TTT GCT GAC CTC AAT His Asn Ile Tyr Ala Glu Tyr Val His Phe Ala Asp Leu Asn 1428 1443 1458 GCG CCG TTA AAC GAT GAT GAG CAC GCA CAA CTT GAA CGC CTG Ala Pro Leu Asn Asp Asp Glu His Ala Gln Leu Glu Arg Leu 1473 1488 1503 CTG AAA TAT GGC CCG GCA CTC GCC AGC CAC GCC CCG CAA GGC Leu Lys Tyr Gly Pro Ala Leu Ala Ser His Ala Pro Gln Gly 1518
ARA CTC CTG GTG ACC CCG CGT CCT GGC ACC ATC TCT CCC
Lys Leu Leu Leu Val Thr Pro Arg Pro Gly Thr 11e Ber Pro

1563
TGG TCT TCG AAA GCG ACC GAT ATT GCC CAT AAC TGC GGG CTA
TTP Ser Ser Lys Ala Thr Asp Ile Ala His Asn Cys Gly Leu 1608

CAA CAG GTA AAC CGC CTT GAG CGC GGC GTT GCT TAC TAT ATA Gln Gln Val Asn Arg Leu Glu Arg Gly Val Ala Tyr Tyr Ile 1638 1653 1668
GAA GCC GGT ACG CTG ACC AAT GAA CAA TGG CAG CAG GTT ACC
Glu Ala Gly Thr Leu Thr Asn Glu Gln Trp Gln Gln Val Thr GCT GAA CTG CAC GAC CGC ATG ATG GAA ACG GTC TTT TTT GCT Ala Glu Leu His Asp Arg MET MET Glu Thr Val Phe Phe Ala 1728 1743 1758
TTA GAT GAT GCA GAG CAG TTG TTT GCC CAC CAT CAA CCG ACT
Leu Asp Aep Ala Glu Gln Leu Phe Ala His His Gln Pro Thr 1773 1788 1803
CCG GTT ACC AGC GTT GAT TTG CTG GGG CAG GGC CGT CAG GCG
Pro Val Thr Ser Val Asp Leu Leu Gly Gln Gly Arg Gln Ala 1818 1833 CTG ATC GAC GCT AAC CTG CGT CTT GGC TTG GCT CTG GCG GAA Leu Ile Asp Ala Asn Leu Arg Leu Gly Leu Ala Leu Ala Glu 1848 1863 1878 GAT GAA ATT GAC TAT CTG CAG GAT GCT TTC ACA AAG CTT GGT Asp Glu Ile Asp Tyr Leu Gln Asp Ala Phe Thr Lys Leu Gly 1893 1908 1923 CGT AAC CCG AAC GAC ATC GAA CTG TAT ATG TTT GCC CAG GCG Arg Asn Pro Asn Asp Ile Glu Leu Tyr MET Phe Ala Gln Ala 1938 1953 1968 ANC TCC GAG CAC TGC CGC CAC ANN ATT TTT ANC GCC GAC TGG Asn Ser Glu His Cys Arg His Lys Ile Phe Asn Ala Asp Trp 1983 1998 2013
GTT ATC GAT GGT GAA CAG CAG CCG AAA TCG CTG TTC AAG ATG
Val 1le Asp Gly Glu Gln Gln Pro Lys Ser Leu Phe Lys MET 2028
ATC AAA AAT ACT TTC GAA ACC ACG CCA GAT CAC GTT CTC TCT
Ile Lys Asn Thr Phe Glu Thr Thr Pro Asp His Val Leu Ser 2073 2088 GCT TAT AAA GAT AAC GCC GCC GTA ATG GAA GGT TCT GAA GTG Ala Tyr Lys Asp Asn Ala Ala Val MET Glu Gly Ser Glu Val 2103

GGC CGC TAC TTT GCT GAC CAC GAA ACG GGC CGC TAC GAT TTC
Gly Arg Tyr Phe Ala Asp His Glu Thr Gly Arg Tyr Asp Phe 2148 2163 2178
CAT CAG GAA CCG GCG CAT ATT CTG ATG AAA GTC GAA ACT CAC
His Gln Glu Pro Ala His Ile Leu MET Lys Val Glu Thr His 2193 2208 2223
AAC CAC CGG AGG GGG ATT TCT CGG TGG CGG GGG GGG ACC
Asn His Pro Thr Ala Ile Ser Pro Trp Pro Gly Ala Ala Thr GGT TCC GGC GGT GAA ATC CGC GAT GAA GGT GCC ACC GGG CGC Gly Ser Gly Gly Glu Ile Arg Asp Glu Gly Ala Thr Gly Arg 2268 2298 GGC GCA AAG CCG AAA GCG GGT CTG GTT GGT TTC TCC GTT TCC Gly Ala Lys Pro Lys Ala Gly Leu Val Gly Phe Ser Val Ser ARC CTG CGA ATT CCT GGC TTC GAR CAG CCG TGG GAA GAA GAT Asn Leu Arg Ile Pro Gly Phe Glu Gln Pro Trp Glu Glu Asp 2358
TTC GGT AAG CCT GAG CGC ATT GTC ACC GCG CTG GAC ATC ATG
Phe Gly Lys Pro Glu Arg Ile Val Thr Ala Leu Asp Ile MET 2403
ACC GAA GGC CCG CTG GGC GGC GCG GCG TTT AAC AAC GAA TTT
Thr Glu Gly Pro Leu Gly Gly Ala Ala Phe Asn Asn Glu Phe 2448

GGT CGT CCG GCA CTG AAC GGC TAC TTC CGT ACT TAT GAA GAA
Gly Arg Pro Ala Leu Asn Gly Tyr Phe Arg Thr Tyr Glu Glu 2478 2493 2508 ARA GTG ARC AGC CAC ARC GGC GAR GRG CTG CGC GGT TAT CRC Lys Val Asn Ser His Asn Gly Glu Glu Leu Arg Gly Tyr His 2523 2538 2553 2553 AAA CCG ATC ATG CTG GCG GGC GGG ATC GGC AAC ATT CGC GCC Lys Pro Ile MET Leu Ala Gly Gly Ile Gly Asn Ile Arg Ala 2568 2583 2598
GAT CAC GTA CAA AAA GGC GAG ATC AAC GTC GGT GCG AAG CTG
ASP His Val Gln Lys Gly Glu Ile Asn Val Gly Ala Lys Leu 2613 2628 2643
GTC GTT CTC GGC GGC CCG GCA ATG AAC ATC GGT CTT GGC GGT
Val Val Leu Gly Gly Pro Ala MET Asn Ile Gly Leu Gly Gly 2658
2673
GGT GCA GCG TCT TCT ATG GCG TCT GGT CAG TCT GAT GCC GAC
Gly Ala Ala ser ser MET Ala ser Gly Gln ser Asp Ala Asp 2668 2703 2718
CTC GAC TTT GCT TCC GTA CAG CGC GAC AAC CCG GAG ATG GAG
Leu Asp Phe Ala Ser Val Gln Arg Asp Asn Pro Glu MET Glu 2733 2748 2763 CGT CGC CAG GAA GTG ATC GAC CGT TGC TGG CAG CTT GGT Arg Arg Cys Gln Glu Val Ile Asp Arg Cys Trp Gln Leu Gly 2778 2793 2808
GAT GCC ARC CCA ATC CTG TTT ATC CAC GAC GTT GGC GCT GGC
ASP Ala Asn Pro Ile Leu Phe Ile His Asp Val Gly Ala Gly 2823 2838 2853
GGT CTT TCT AAC GCC ATG CCG GAA CTG GTG AGC GAC GGC GGG
Gly Leu Ser Asn Ala MET Pro Glu Leu Val Ser Asp Gly Gly 2869 2883
CGC GGC GGT AAA TTT GAA CTG CGC GAG ATT CTA AGC GAC GAA
Arg Gly Gly Lys Phe Glu Leu Arg Glu Ile Leu Ser Asp Glu 2913 2928 CCG GGC ATG AGC CCG CTG GAA ATC TGG TGT AAC GAA TCC CAG Pro Gly MET Ser Pro Leu Glu Ile Trp Cys Asn Glu Ser Gln 2943
GAA CGC TAC GTG CTG GCG GTT GCT GCC GAT CAA TTA CCG CTG
Glu Arg Tyr Val Leu Ala Val Ala Ala Asp Gln Leu Pro Leu 2988 3003 3018
TTT GAC GAA CTG TGT AAG CGT GAG CGC GCA CCC TAC GCG GTG
Phe Asp Glu Leu Cys Lys Arg Glu Arg Ala Pro Tyr Ala Val 3033 3048 3063 ATT GGT GAA GCG ACC GAA GAA CTG CAT CTT TCT CTG CAC GAT Ile Gly Glu Ala Thr Glu Glu Leu His Leu Ser Leu His Asp CGT CAT TTT GAT AAT CAG CCG ATC GAT CTG CCG CTG GAC GTC Arg His Phe Asp Asn Gln Pro Ile Asp Leu Pro Leu Asp Val 3108 3123 3138 CTG CTT GGT AAA ACG CCG AAG ATG ACC CGC GAT GTA CAA ACG Leu Leu Gly Lys Thr Pro Lys MET Thr Arg Asp Val Gln Thr 3153 3168 3183 CTG AAA GCG AAA GGC GAC GCG CTG GCC CGT GAA GGG ATC ACC Leu Lys Ala Lys Gly Asp Ala Leu Ala Arg Glu Gly Ile Thr 3198 3213 3228
ATT GCA GAC GCG GTG AAA CGT GTG CTG CAT CTG CCG ACT GTG
Ile Ala Asp Ala Val Lys Arg Val Leu His Leu Pro Thr Val GCG GAA AAA ACC TTC CTG GTG ACC ATT GGC GAC CGC AGC GTA Ala Glu Lys Thr Phe Leu Val Thr Ile Gly Asp Arg Ser Val 3288 3303 ACC GGC ATG GTA GCG CGC GAT CAG ATG GTG GGG CCG TGG CAG Thr Gly MET Val Ala Arg Asp Gln MET Val Gly Pro Trp Gln 3333 3348 GTG CCG GTC GCT AAC TGC GCG GTC ACT ACC GCC AGC CTC GAC Val Pro Val Ala Asn Cys Ala Val Thr Thr Ala Ser Leu Asp 3363 3378 3393
AGC TAC TAC GGT GAA GCG ATG GCG ATT GGC GAG CGT GCG CCG
Ser Tyr Tyr Gly Glu Ala MET Ala Ile Gly Glu Arg Ala Pro 3408 3423 3438 GTT GCG CTG CTG GAT TTC GCC GCC TCT GCC CGT CTG GCG GTC Val Ala Leu Leu Asp Phe Ala Ala Ser Ala Arg Leu Ala Val 3498 3513
ATC ARA CGC ATC ARA CTT TCC GCC ARC TGG ATG GCG GCG GCA
Ile Lys Arg Ile Lys Leu Ser Ala Asn Trp MET Ala Ala Ala 3528 3543 3558

GGC CAC CCT GGT GAA GAT GCG GGC CTG TAT GAA GCC GTT AAA
Gly His Pro Gly Glu Asp Ala Gly Leu Tyr Glu Ala Val Lys 3573 3588 3603 GCC GTG GGC GAA GAG CTT TGT CCG GCG CTG GGC CTG ACG ATC Ala Val Gly Glu Glu Leu Cys Pro Ala Leu Gly Leu Thr Ile . 3618 3633 3648 CCG GTG GGT ARA GAC TCC ATG TCG ATG ARA ACC CGC TGG CAG Pro Val Gly Lys Asp Ser MET Ser MET Lys Thr Arg Trp Gln 3663 3678 3693
GAA GGT AAC GAA GAG CGC GAA ATG ACG TCG CCG CTG TCG CTG
Glu Gly Asn Glu Glu Arg Glu MET Thr Ser Pro Leu Ser Leu 3708 3723 GTG ATT TCT GCA TTT GCC CGC GTG GAA GAT GTA CGT CAC ACC Val lle Ser Ala Phe Ala Arg Val Glu Asp Val Arg His Thr 3738 3753 3768
ATC ACG CCG CAG CTT TCT ACC GAA GAT AAC GCA CTG CTG CTG
Ile Thr Pro Gln Leu Ser Thr Glu Asp Asn Ala Leu Leu Leu ATT GAT TTG GGC AAA GGC AAT AAC GCG CTG GGC GCA ACG GCG Ile Asp Leu Gly Lys Gly Asn Asn Ala Leu Gly Ala Thr Ala

CTG GCG CAG GTT TAT CGT CAG CTT GGC GAC AAA CCG GCA GAT Leu ala Gln Val Tyr Arg Gln Leu Gly Asp Lys Pro Ala Asp GTA CGC GAT GTC GCG CAA CTG AAA GGC TTC TAT GAC GCG ATT Val Arg Asp Val Ala Gln Leu Lys Gly Phe Tyr Asp Ala Ile 3918 3933 CAG GCG CTG GTT GCA CAG CGT AAG CTG CTG GCG TAT CAC GAC Gln Ala Leu Val Ala Gln Arg Lys Leu Leu Ala Tyr His Asp CGC TCT GAT GGC GGC CTG CTG GTA ACG CTG GCG GAA ATG GCC Arg Ser Asp Gly Gly Leu Leu Val Thr Leu Ala Glu MET Ala 3993 4008 4023
TTT GCT GGT CAT TGT GGC ATT GAC GCG GAT ATC GCC ACT CTG
Phe Ala Gly His Cys Gly Ile Asp Ala Asp Ile Ala Thr Leu GGT GAC GAT CGC CTG GCG GCG TTG TTT AAC GAA GAA CTG GGT Gly Asp Asp Arg Leu Ala Ala Leu Phe Asn Glu Glu Leu Gly 4083 4098 4113
GCG GTG ATT CAG GTT CGT GCC GCT GAC CGT GAA GCG GTC GAG
Ala Val Ile Gln Val Arg Ala Ala Asp Arg Glu Ala Val Glu TCC GTA CTG GCA CAG CAT GGG CTT GCT GAT TGT GTC CAT TAT Ser Val Leu Ala Gln Eis Gly Leu Ala Asp Cys Val Eis Tyr 4158 4173 4188
GTA GGG CAG GCG GTT TCC GGT GAC CGT TTT GTG ATT ACC GCC
Val Gly Gln Ala Val Ser Gly Asp Arg Phe Val Ile Thr Ala 4203
AAC GGG CAG ACT GTA TTC AGC GAA AGC CGC ACC ACG TTG CGT
Asn Gly Gln Thr Val Phe Ser Glu Ser Arg Thr Thr Leu Arg 4248 4263 4278
GTC TGG TGG GCA GAA ACT ACC TGG CAG ATG CAG CGC CTG CGT
Val Trp Trp Ala Glu Thr Thr Trp Gln MET Gln Arg Leu Arg GAC AAC CCG GAG TGT GCC GAT CAG GAG CAT CAG GCG AAA TCT Asp Asn Pro Glu Cys Ala Asp Gln Glu His Gln Ala Lys Ser 4338 4353 ANC GAC GCC GAT CCG GGC CTG ANT GTA ANA CTG TCG TTC GAT Asn Asp Ala Asp Pro Gly Leu Asn Val Lys Leu Ser Phe Asp ATC AAC GAA GAT GTG GCA GCA CCG TAT ATT GCC ACT GGC GCA Ile Asn Glu Asp Val Ala Ala Pro Tyr Ile Ala Thr Gly Ala 4413
CGT CCG ARA GTT GCC GTA CTG CGT GAG CAG GGC GTG ARC TCG
Arg Pro Lys Val Ala Val Leu Arg Glu Gln Gly Val Asn Ser 4458
CAT GTT GAA ATG GCG GCA GCT TTC CAC CGT GCA GGC TTT GAT
His Val Glu MET Ala Ala Ala Phe His Arg Ala Gly Phe Asp 4503 4518 4533 GCT ATC GAC GTG CAT ATG AGT GAC CTG CTG ACC GGA CGC ACG Ala Ile Asp Val His MET Ser Asp Leu Leu Thr Gly Arg Thr 4548 4563
GGC CTG GAA GAT TTC CAC GCC CTG GTC GCG TGC GGT GGT TTC
Gly Leu Glu Asp Phe His Ala Leu Val Ala Cys Gly Gly Phe 4578
4593
TCC TAC GGT GAT GTG CTG GGT GCC GGT GAA GGT TGG GCG AAG
Ber Tyr Gly Asp Val Leu Gly Ala Gly Glu Gly Trp Ala Lys 4623
TCA ATC CTG TTC AAT GAC CGT GTA CGC GAT GAG TTT GCA ACC
Ser Ile Leu Phe Asn Asp Arg Yal Arg Asp Glu Phe Ala Thr TTC TTC CAC CGT CCG CAA ACG CTG GCG CTG GGG GTA TGT AAC
Phe Phe His Arg Pro Gln Thr Leu Ala Leu Gly Val Cys Asn 4713

GGT TGC CAG ATG ATG TCT AAT CTG CGT GAA CTG ATC CCA GGT
Gly Cys Gln MET MET Ser Asn Leu Arg Glu Leu Ile Pro Gly 4758
AGT GAG TTG TGG CCA CGT TTT GTG CGC AAT ACC TCC GAT CGC
Ser Glu Leu Trp Pro Arg Phe Val Arg Asn Thr Ser Asp Arg 4803 TTT GAA GCG CGT TTC AGC CTG GTT GAA GTA ACC CAA AGC CCG Phe Glu Ala Arg Phe Ser Leu Val Glu Val Thr Gln Ser Pro 4833
TCT CTG CTG TTG CAG GGG ATG GTG GGC TCG CAA ATG CCG ATT
Ser Leu Leu Gln Gly MET Val Gly Ser Gln MET Pro Ile 4878
GCT GTC TCT CAT GGT GAA GGG CGC GTG GAA GTG CGT GAT GCG
Ala Val Ser His Gly Glu Gly Arg Val Glu Val Arg Asp Ala GCG CAT CTG GCG GCA CTG GAA AGC AAA GGG CTG GTG GCA CTG Ala His Leu Ala Ala Leu Glu Ser Lys Gly Leu Val Ala Leu

TTAGCGACAC AGTGGCTGAT TTACATTCAA ACTGCGGAGG 5451 AGACATAAAC CCCTCAAAGC AAGCTATAAT CGATAGCTTA CCAAACTATT 5471 5481 5491 5501 TTCTTTATTG GCACAGTTAC TGCATAATAG TAACCAGTGG CTCATTCACC GACTTATGTC AGCCCCTTCG GGACGTGCTA CATAAAATAC GAATGACGCA 5571 5591 5581 5601 CCTGCCGTCC AACTTCTGAT ATCAGCGTAG CTATATCAAC 5631 5641 5651 5661 ACGTCGAGT TAGGCACCGC CTTATTCCAT AACAAAGCCG 5621 CATCGGGCGA GGTAATTCCC GGCTTTGTTG TATCTGAACT TCCCCTCGGT TAGCATCAGG 5721 5731 5741 5751 5761 TAACACCTTG 5771 5791 CCGCTCATTA CGACAACTGG TAATGCTGGC ATTTTTGCTG ATTCTGCTGC 5831 5821 5841 5851 CCCTGTTGGT GCTGGCATGG CAAGCCTGGC AAAGCCTGAA TGCGCTTAGC

FIGURE 3: Nucleotide and deduced amino acid sequence of purL. The DNA sequence of the sense strand of the purL gene is shown. It is numbered from a Sau3A site as the 5' end and includes 1293 nucleotides of upstream DNA, 3885 nucleotides of coding region, and 687 nucleotides beyond the coding region. The proposed purL Shine-Dalgarno sequence (Shine & Dalgarno, 1974) at nucleotides 1285-1287 is indicated as S/D. The amino acid residues confirmed by sequencing are overlined. The FGAM synthetase sequence matching the active site sequence of the Salmonella and chicken liver FGAM synthetase is also overlined. At 58 bp downstream from the purL structural gene is a region of dyad symmetry (nucleotides 5240-5261) with the characteristics of a  $\rho$  independent terminator (Holmes et al., 1983), a  $\Delta G$  of -18.8 kcal/mol (Tinoco et al., 1973), and that could represent the termination site of purL transcription. Nucleotide 1248, identified by primer extension studies, is marked with an asterisk. The inferred -10 and -35 regions are underlined and labeled. The common sequence (nucleotides 1204-1216) shared with the other pur loci is also underlined.

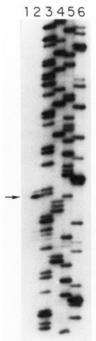


FIGURE 4: Primer extension analysis of the 5' end of purL mRNA. An oligonucleotide complementary to nucleotides 1324–1343 within the purL coding region was hybridized to total cellular RNA from strain TX540 containing plasmid pJS80. It was extended with reverse transcriptase, and the products were analyzed on a sequencing gel. Lane 1 is a control reaction lacking the 20-mer oligonucleotide, lane 2 is the primer extension product, and lanes 3-6 are the dideoxy sequencing reactions used as size standards. The major primer extension product is 96 nucleotides in length.

Physical Characterization of FGAM Synthetase. SDS gel electrophoresis (Figure 5) and Sephadex G-150 chromatography of E. coli FGAM synthetase both indicate by comparison with standards of known molecular weight that FGAM synthetase is a single polypeptide of  $M_r = 135$ K and 125K, respectively. These numbers are similar to the previously

step <sup>a</sup>	of FGAM vol (mL)	protein (mg/mL)	total act.	sp act. (units mg <sup>-1</sup> )
crude	5.2	14.4	81	1.08
40-55% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	0.83	57.0	59	1.25
Sephadex G-25	31	1.4	64	1.47
DEAE-52	145	0.16	50	2.15

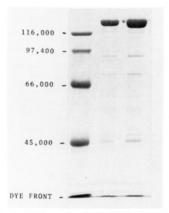


FIGURE 5: SDS gel (7.5% acrylamide) of E. coli FGAM synthetase: lane 1, molecular weight standards; lane 2, 20  $\mu$ g of FGAM synthetase from the DEAE-52 column (Table I); lane 3, 40  $\mu$ g.

determined molecular weight of the chicken liver (133K) and Salmonella (135K) protein and are slightly less than the  $M_r$  = 141K determined by the  $E.\ coli$  gene sequence.

Catalytic Properties of FGAM Synthetase. E. coli FGAM synthetase has been shown to catalyze the stoichiometric conversion of FGAR, MgATP, and glutamine to FGAM, MgADP, and glutamate with the assay procedures described under Materials and Methods.

These studies are a prerequisite to the investigation of whether E. coli FGAM synthetase, in analogy with all other amidotransferase thus far examined, exhibits glutaminase

Table II: Comparison of Conserved Regions of Glutamine Amide Transfer<sup>a</sup>

-		
Enzyme (Source)	ref.	
FGAMS (E. Coli)	NH <sub>2</sub> 1120T L A L G V C N G C Q M M121S N S W H P E20CO <sub>2</sub> H <sup>1</sup>	
FGAMS (S. typhimurium)	ALGVC CO <sub>2</sub> H <sup>2</sup>	
FGAMS (B. subtilis)	NH <sub>2</sub> 79K P V L G V C N G F Q I L97G M M P H P E31CO <sub>2</sub> H <sup>3</sup>	
CTPS (E. Coli)	NH <sub>2</sub> 72I P Y L G I C L G M Q V A125A C Q F H P E28CO <sub>2</sub> H <sup>4</sup>	
CPS (E. Coli)	NH <sub>2</sub> 263I P V F G I C L G H Q L L73S F Q G H P E27CO <sub>2</sub> H <sup>5</sup>	
PABS (E. Coli)	NH <sub>2</sub> 72L P I L G V C L G H Q A M78G V Q F H P E17CO <sub>2</sub> H <sup>6</sup>	
ASII (E. Coli)	NH <sub>2</sub> 80L P I I G I C L G H Q A I75G F Q F H P E33CO <sub>2</sub> H <sup>7</sup>	
GMPS (E. Coli)	NH <sub>2</sub> 79V P V F G V C Y G M Q T M84G V O F H P E23CO <sub>2</sub> H <sup>8</sup>	

<sup>a</sup>References: (1) this paper; (2) Dawid et al., 1963; (3) Ebbole & Zalkin, 1987; (4) Weng et al., 1986; (5) Piette et al., 1984; (6) Kaplan et al., 1985; (7) Nichols et al., 1980; (8) Zalkin et al., 1985.

activity and the ability to use NH3 in place of glutamine. Incubation of glutamine with E. coli FGAM synthetase at pH 7.3 (optimum) resulted in the production of glutamate at 0.02% the rate of the normal reaction. This result contrasts with previously reported glutaminase activities of 0.5% for the chicken liver (Mizobuchi & Buchanan, 1968a) and 0.6% for the Ehrlich ascites tumor cell enzyme (Chu & Henderson, 1972). A number of glutamine-requiring enzymes exhibit enhanced glutaminase activity in the presence of one of their additional substrates, or in the presence of substrate analogues. The glutaminase activity of FGAM synthetase was therefore determined in the presence of FGAR, MgATP, or FGAR and MgAMP-PCP. The glutaminase activity was unaffected by any of these alternative conditions. Previous studies by Mizobuchi and Buchanan (1968a) showed that incubation of the chicken liver enzyme with glutamine and NH2OH resulted in the production of the  $\gamma$ -glutamylhydroxamate at a rate of 80% that of the conversion of FGAR to FGAM. A similar experiment using the E. coli enzyme failed to reveal production of any detectable hydroxamate. Thus the glutamate binding domains of the chicken liver and E. coli synthetases, even though their active site cysteines are situated in similar environments (Table II), possess marked differences in accessibility to small molecules such as H<sub>2</sub>O and NH<sub>2</sub>OH.

As in the case of other amidotransferases,  $E.\ coli\ FGAM$  synthetase is also capable of using ammonia in place of glutamine as the nitrogen donor in the reaction. However, the  $K_m$  for ammonium chloride is 1 M at pH 8.0 and the  $V_{max}$  is only 2% that of the glutamine reaction. These results again contrast with those reported for the chicken liver synthetase by Mizobuchi and Buchanan (1968):  $K_m$  of 10 mM and a  $V_{max}$  of 5% the normal turnover number.

Characterization of the FGAM Synthetase-Glutamine Complex. Chicken liver FGAM synthetase (Mizobuchi & Buchanan, 1968a), E. coli CTP synthetase (Levitzki & Koshland, 1971), and E. coli carbamoylphosphate synthetase (Wellner et al., 1973) have in common the ability to form an isolable 1:1 complex with glutamine. This subgroup of amidotransferases has a homologous trpG-derived glutamine amide transfer domain containing both an invariant and catalytically essential cysteine and imidazole (Table II) (Ebbole & Zalkin, 1987). The nature of the complex formed by

CTP synthetase and chicken liver FGAM synthetase was proposed to consist of a covalent  $\gamma$ -glutamyl enzyme thioester. The complex formed by carbamoylphosphate synthetase was proposed to consist of a tightly bound glutamine (Wellner et al., 1973; Rubino et al., 1986). A similar series of experiments were undertaken to determine the composition of a complex, if formed, between  $E.\ coli$  FGAM synthetase and glutamine. Incubation of  $[U^{-14}C]$ -L-glutamine with  $E.\ coli$  FGAM synthetase for 5 min, followed by chromatography on Sephadex G-50, resulted in coelution of radioactivity with the protein. A control experiment in which  $[U^{-14}C]$  glutamate replaced glutamine indicated no coelution of radioactivity with protein.

The half-life for the complex is 22 min at 4 °C, determined from a semilog plot of the binding ratio (glutamine:enzyme) as a function of time. Extrapolation to zero time gives a binding ratio of 0.96. Thus FGAM synthetase from E. coli is also capable of binding 1 equiv of glutamine per monomer.

The nature of this complex was investigated by determining the composition of the radioactive material which coeluted with the enzyme. When the isolated complex was allowed to stand at 4 °C for 4 h or was denatured by the addition of 20% trichloroacetic acid, the radioactivity was all found to comigrate with glutamate on an anion-exchange column. A control experiment in which glutamine was treated in an identical manner gave no detectable hydrolysis to glutamate. This result suggested that the glutamine may be bound as a thiohemiaminal or as a thioester in the complex and that the intermediate(s) can hydrolyze to form glutamate. The slow breakdown of the complex to glutamate is consistent with the slow rate of glutaminase activity catalyzed by the enzyme. To determine if glutamine is covalently bound as a thioester to FGAM synthetase, the isolated complex was treated with hydroxylamine at pH 7.2 for 20 min. Hydroxylamine is known to react quantitatively with thioesters to form hydroxamic acids (Stadtman, 1957). As shown in Table III, no glutamylhydroxamic acid was detectable, and 95% of the radioactivity was recovered as glutamate. Inaccessibility of hydroxylamine to the active site provides a possible explanation of these observations.

In order to enhance the accessibility of hydroxylamine to the putative thioester, the hydroxylamine trapping experiment

Table III: Trapping of Enzyme-Glutamine Complex with Hydroxylamine

complex treatment	enzyme (nmol)	complex (nmol)	glutamate (nmol)	glutamine or γ-glutamyl- hydroxamic acid (nmol)
1 M hydroxylamine	6.0	4.0	3.8	undetectable
1 M hydroxylamine with 6 M guanidine hydrochloride	5.0	3.4	0.45	2.8

was repeated in the presence of 6 M guanidine hydrochloride at pH 7.2. Isolation of the products from this experiment established that now 85% of the radioactivity comigrated with γ-glutamylhydroxamic acid, while 15% comigrated with glutamate on an anion-exchange column (Table III). Since glutamine and  $\gamma$ -glutamylhydroxamic acid coelute by this chromatographic procedure, the method of Levintow et al. (1955) was utilized to distinguish between these two alternatives. Previous investigators have established and we have substantiated that under a defined set of conditions that  $\gamma$ glutamylhydroxamic acid can be quantitatively converted to pyrrolidonecarboxylic acid while only 4% of the glutamine and none of the glutamate treated in the same manner are converted to this product. Therefore, our sample was heated at 100 °C for 7.5 min at neutral pH and rechromatographed on an anion-exchange column. The eluents were then chromatographed on a cellulose F thin-layer plate (Materials and Methods) which has been shown to separate L-glutamate, L-glutamine, and pyrrolidonecarboxylic acid (Mizobuchi & Buchanan, 1968a). Location of the radioactive material by scintillation counting revealed that it all comigrated with pyrrolidonecarboxylic acid. These experiments show that, as in the case of the chicken liver FGAM synthetase (Mizobuchi & Buchanan, 1968a) and E. coli cytidine-5'-triphosphate synthetase (Levitzki & Koshland, 1971), the enzyme-glutamine complex consists of a covalent  $\gamma$ -glutamyl enzyme thioester or the tetrahedral precursor to the thiol ester which is capable of undergoing hydrolysis to glutamate.

Having established that the enzyme-glutamine complex consists of a  $\gamma$ -glutamyl enzyme thioester, a question remains concerning the location of the "putative ammonia equivalent". Several possibilities exist: the ammonia may be held tightly (covalently or noncovalently) at the active site of FGAM synthetase ready to undergo reaction when MgATP and FGAR are added, or it may have dissociated from the complex. To determine if the enzyme-glutamine complex is chemically competent to produce products, the following experiment was carried out. The enzyme-glutamine complex was isolated from a Sephadex G-50 column and immediately added to a reaction mixture containing FGAR, MgATP, and AIR synthetase. If any FGAM is produced, AIR synthetase will convert it to AIR which can be quantitated by the Bratton-Marshall assay. When 6.8 nmol of complex is treated in this fashion, no AIR is formed (Table IV). A control to set a lower limit on product detection was run in which glutamine (2.1 or 4.2 nmol) was included in an identical reaction mixture. The expected amount of AIR was produced (Table IV). This experiment demonstrates that the enzyme-glutamine complex is not chemically competent for product formation in the absence of added glutamine. Hence, at least 99% of the ammonia liberated by thioester formation either remains bound in a chemically incompetent fashion or more probably has dissociated from the active site. These observations seem to rule

Table IV: Turnover of Enzyme-Glutamine Complex To Form

substrates added to E-Gln complex	enzyme (nmol)	complex (nmol)	AIR (nmol)
MgATP, FGAR	10	6.8	undetectable
MgATP, FGAR, 2.1 nmol of Gln	7.3	4.8	2.0
MgATP, FGAR, 4.2 nmol of Gln	6.6	4.6	4.3

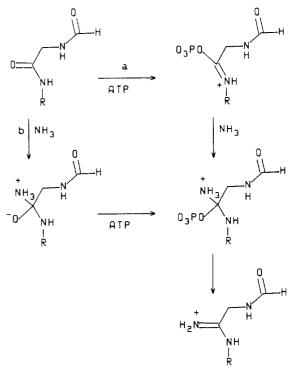


FIGURE 6: Proposed mechanism for the conversion of FGAR to FGAM catalyzed by FGAM synthetase. R = ribose 5-phosphate.

out the chemically attractive possibility that the putative ammonia is held covalently at the active site as a thiohemiaminal ready to deliver NH<sub>3</sub> on binding of FGAR and ATP.

Transfer of <sup>18</sup>O from [<sup>18</sup>O]FGAR to P<sub>i</sub>. Several mechanisms for the conversion of FGAR and FGAM have been proposed for the chicken liver FGAM synthetase (Schendel & Stubbe, 1986) (Figure 6). Pathway a (Figure 6) originally suggested by Satterthwait and Westheimer (1980) involves phosphorylation of the amide oxygen of FGAR by ATP to form a phosphorylated FGAR intermediate. This species is then attacked by putative NH3 produced from glutamine to form a tetrahedral intermediate which collapses with subsequent loss of phosphate to form FGAM. Additional support for this type of mechanism comes from the recent work by von der Saal et al. (1985) investigating cytidine-5'-triphosphate synthetase using the positional isotope exchange methods. Pathway b, a modification of a mechanism proposed by Mizobuchi et al. (1968), involves attack on the carbonyl of the amide by ammonia to form a carbinolamine which is then phosphorylated by ATP, followed by loss of phosphate to give FGAM. A similar mechanism in which the attack by ammonia and phosphorylation by ATP occur in a concerted fashion is also possible. In addition or alternatively, a phosphorylated enzyme intermediate must be considered.

Both mechanisms in Figure 6 predict that the oxygen from the amide of FGAR would be transferred to the inorganic phosphate derived from ATP hydrolysis. To test this hypothesis, FGAR was prepared with 50% <sup>18</sup>O in the amide

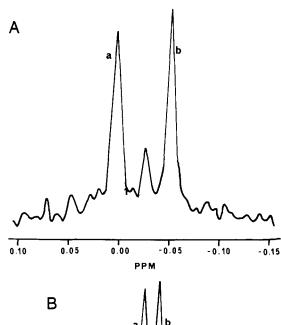
substrate	constant substrates	inhibitor	inhibition type	$K_{is}$ (mM)	$K_{ii}$ (mM)
Gln	FGAR (72 μM), ATP (106 μM)	glutamate	С	$1.6 \pm 0.16$	.=
FGAR	Gln (88 $\mu$ M), ATP (106 $\mu$ M)	glutamate	NC	$13 \pm 1.8$	$52 \pm 9.9$
ATP	FGAR (72 $\mu$ M), Gln (88 $\mu$ M)	glutamate	NC	$21 \pm 5.3$	$16 \pm 2.1$
Gln	FGAR (89 $\mu$ M), ATP (101 $\mu$ M)	albizziin	С	$8.9 \pm 0.69$	
FGAR	Gln (110 $\mu$ M), ATP (139 $\mu$ M)	albizziin	NC	$15 \pm 4.7$	$23 \pm 6.2$
ATP	FGAR (89 $\mu$ M), Gln (110 $\mu$ M)	albizziin	NC	$13 \pm 2.0$	$18 \pm 1.6$
ATP	FGAR (72 $\mu$ M), Gln (138 $\mu$ M)	AMP-PCP <sup>a</sup>	C	$0.066 \pm 0.005$	
FGAR	ATP (134 $\mu$ M), Gln (138 $\mu$ M)	AMP-PCP	NC	$2.0 \pm 0.37$	$6.4 \pm 1.8$
Gln	FGAR (72 $\mu$ M), ATP (134 $\mu$ M)	AMP-PCP	UC		$2.2 \pm 0.10$

<sup>a</sup> Adenosine 5'-( $\beta$ , $\gamma$ -methylenetriphosphate).

oxygen. This was accomplished by mixing equal amounts of  $[1-{}^{16}O_2, {}^{13}C]$  glycine and  $[1-{}^{18}O_2, {}^{13}C]$  glycine (Figure 7A) and biosynthetically incorporating the glycine into FGAR (Materials and Methods). The [18O/16O]FGAR is incubated with FGAM synthetase, MgATP, and glutamine, and the inorganic phosphate which is produced is isolated by DEAE-Sephadex anion-exchange chromatography. A 31P NMR spectrum of the isolated inorganic phosphate was taken and is shown in Figure 7B. This spectrum indicates that a mixture of [16O]P<sub>i</sub> and [18O]P<sub>i</sub> is formed in a ratio which is identical with the <sup>16</sup>O/<sup>18</sup>O ratio in the starting glycine. Therefore, the amide oxygen of FGAR is quantitatively transferred to the phosphate derived from ATP hydrolysis during the conversion of FGAR to FGAM.

Kinetic Studies. In order to determine the optimum conditions for kinetic analysis, the pH dependence and metal ion specificity of FGAM synthetase were investigated. The enzyme has a pH optimum of 7.2 with only 50% of this activity being retained at pH 6.2 or 8.0. In addition, 20 mM free Mg<sup>2+</sup> and 10 mM free K+ ions were found to be required for optimal activity. The requirement for high concentrations of free Mg<sup>2+</sup> suggest a structural role for the metal in addition to its substrate role as MgATP. These results can be compared with results from the chicken liver enzyme which has a pH optimum of 8 and requires 60 mM free K+ and 20 mM free Mg2+ for optimal activity (Mizobuchi & Buchanan, 1968b). Determination of the Michaelis constants for the three substrates resulted in a  $K_{\rm m}=64~\mu{\rm M}$  for glutamine, a  $K_{\rm m}=51~\mu{\rm M}$  for MgATP, and a  $K_{\rm m}=30~\mu{\rm M}$  for FGAR. The  $K_{\rm m}$  for MgATP is considerably different from the  $K_m$  of 1.5 mM reported for both the chicken liver and tumor cell enzymes, while the  $K_{\rm m}$ for glutamine and FGAR are similar, 40  $\mu$ M and 100  $\mu$ M (chicken liver) and 110  $\mu$ M and 110  $\mu$ M (Erlich ascites tumor cells), respectively (Li & Buchanan, 1971; Chu & Henderson, 1972).

Initial Velocity and Product and Dead-End Inhibition Studies. Initial velocity and product and dead-end inhibition studies were undertaken to determine the order of addition of substrates in this three-substrate, four-product system. The initial velocity studies resulted in an intersecting pattern when FGAR and MgATP were varied at fixed levels of glutamine. However, when FGAR and glutamine were varied at fixed levels of MgATP, or when MgATP and glutamine were varied at fixed levels of FGAR, parallel patterns were obtained. These parallel initial velocity patterns suggest a ping-pong mechanism in which an irreversible step, presumably product release, occurs between the binding of glutamine and the binding of the other two substrates. The intersecting initial velocity pattern between FGAR and MgATP suggests a sequential mechanism for the binding of these two substrates. Two ping-pong mechanisms exist which are consistent with these initial velocity patterns: (1) A mechanism in which glutamine binds first followed by release of glutamate and then



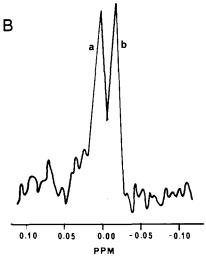


FIGURE 7: (A) A 50.3-MHz <sup>13</sup>C NMR spectrum of a mixture of [1-<sup>13</sup>C, <sup>18</sup>O<sub>2</sub>]glycine and [1-<sup>13</sup>C, <sup>16</sup>O<sub>2</sub>]glycine. Peak a is [1-<sup>13</sup>C, <sup>16</sup>O<sub>2</sub>]glycine and peak b is [1-<sup>13</sup>C, <sup>18</sup>O<sub>2</sub>]glycine. The peak in the center is due to [1-<sup>13</sup>C, <sup>16</sup>O, <sup>18</sup>O]glycine. The spectrum was obtained from 100 acquisitions using broad-band proton decoupling with a 35° pulse angle, a 4K data block, a ±200-Hz sweep width, a 10-s acquisition time, and a 10-s pulse delay. A line-broadening factor of 0.2 Hz was applied to the FID before Fourier transformation. (B) An 80.9-MHz <sup>31</sup>P NMR spectrum showing the inorganic phosphate produced from the reaction of FGAM synthetase, MgATP, glutamine, and a 1:1 mixture of [18O]FGAR and [16O]FGAR. Peak a is [16O]FGAR, and peak b is [18O]FGAR.

the addition of MgATP and FGAR followed by reaction to form products or (2) a mechanism in which MgATP and FGAR bind followed by an irreversible step, possibly the release of ADP, and then the addition of glutamine followed by reaction to form products. However, both mechanisms 1 and 2 are unappealing because of the absence of glutamine/glutamate exchange (mechanism 1) and the slow rate of ATP/ADP exchange (mechanism 2).

To determine which, if either, of these ping-pong mechanisms pertains, product and dead-end inhibition studies were carried out. As shown in Table V, glutamate was found to be competitive with glutamine and noncompetitive with both FGAR and MgATP. This rules out mechanism 1 since this mechanism predicts that glutamate would show a competitive inhibition pattern with either FGAR or MgATP. When albizziin, a glutamine analogue, was used as an inhibitor, it was found to be competitive with glutamine and noncompetitive with both FGAR and MgATP. These data also suggest that the reaction does not proceed by mechanism 1 since this mechanism would predict that albizziin would be uncompetitive with both FGAR and MgATP. These inhibition studies and preliminary exchange studies have therefore caused us to reevaluate our interpretation of our initial velocity studies. Since parallel initial velocity patterns are often difficult to distinguish from patterns which intersect at some point a great distance from the y axis, alternative mechanisms were sought by computer fitting of the initial velocity data to eq 3 and by graphical analysis methods described by Viola and Cleland (1982). From these analyses, the terms corresponding to coeff B, coeff C, and  $K_b$  in eq 3 were found to be undefined. Therefore, the initial velocity data were refit to a form of eq 3 which had these three terms removed. This form of the equation gave the best fit to the initial velocity data and corresponds to the ordered mechanisms shown in Figure 8 (Viola & Cleland, 1982), in which glutamine adds first followed by rapid equilibrium addition of MgATP and then followed by addition of FGAR. This mechanism is consistent with the initial velocity, dead-end inhibition, and substrate/ product exchange studies.

The results of these studies differ from the kinetic mechanism for the chicken liver enzyme where glutamine adds first followed by random addition of MgATP and FGAR (Li & Buchanan, 1971) and from the proposed kinetic mechanism for the Erlich ascites tumor cell enzyme in which glutamine adds first followed by release of glutamate, MgATP adds next followed by release of ADP, and FGAR adds last followed by release of FGAM and P<sub>i</sub> (Chu & Henderson, 1972).

The ability to reproducibly isolate large amounts of FGAM synthetase, the availability of sequence information, and the preliminary mechanistic studies reported herein provide the framework for future detailed mechanistic and regulatory studies on this intriguing amidotransferase.

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## Drosophila Insulin Degrading Enzyme and Rat Skeletal Muscle Insulin Protease Cleave Insulin at Similar Sites<sup>†</sup>

W. C. Duckworth,\*,<sup>‡</sup> J. V. Garcia,<sup>§,</sup> J. J. Liepnieks, <sup>⊥</sup> F. G. Hamel, <sup>‡</sup> M. A. Hermodson, <sup>#</sup> B. H. Frank, <sup>o</sup> and Marsha Rich Rosner<sup>‡, o</sup>

Veterans Administration Medical Center and University of Nebraska Medical Center, Omaha, Nebraska 68105, Massachusetts
Institute of Technology, Cambridge, Massachusetts 02139, Veterans Administration Medical Center,
Indianapolis, Indiana 46285, Purdue University, West Lafayette, Indiana 47907, and Eli Lilly Research Laboratories,
Indianapolis, Indiana 46285

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ABSTRACT: Insulin degradation is an integral part of the cellular action of insulin. Recent evidence suggests that the enzyme insulin protease is involved in the degradation of insulin in mammalian tissues. Drosophila, which has insulin-like hormones and insulin receptor homologues, also expresses an insulin degrading enzyme with properties that are very similar to those of mammalian insulin protease. In the present study, the insulin cleavage products generated by the Drosophila insulin degrading enzyme were identified and compared with the products generated by the mammalian insulin protease. Both purified enzymes were incubated with porcine insulin specifically labeled with <sup>125</sup>I on either the A19 or B26 position, and the degradation products were analyzed by HPLC before and after sulfitolysis. Isolation and sequencing of the cleavage products indicated that both enzymes cleave the A chain of intact insulin at identical sites between residues A13 and A14 and A14 and A15. Sequencing of the B chain fragments demonstrated that the Drosophila enzyme cleaves the B chain of insulin at four sites between residues B10 and B11, B14 and B15, B16 and B17, and B25 and B26. These cleavage sites correspond to four of the seven cleavage sites generated by the mammalian insulin protease. These results demonstrate that all the insulin cleavage sites generated by the Drosopohila insulin degrading enzyme are shared in common with the mammalian insulin protease. These data support the hypothesis that there is evolutionary conservation of the insulin degrading enzyme and further suggest that this enzyme plays an important role in cellular function.

Insulin is an important modulator of cellular growth and metabolism. The mechanisms by which insulin exerts its effects on cells are as yet incompletely understood, but certain components of the system have been characterized. The insulin

Massachusetts Institute of Technology.

signaling system includes the hormone, a specific receptor on the responding cell membrane, a mechanism for signal transmission that may involve intracellular messengers, and a degradative process for removing and inactivating the hormone.

Both the biological response and degradation of the hormone are initiated by binding of insulin to a specific membrane receptor (Terris & Steiner, 1975). The insulin receptor consists of two subunits, an  $\alpha$  subunit which contains the primary binding site and a  $\beta$  subunit which has tyrosine kinase activity and appears to be important in propagating the intracellular signal (Jacobs et al., 1979; Massague et al., 1980; Kasuga et al., 1981; Roth & Cassel, 1983). Subsequent to binding, insulin is internalized via the endosomal pathway and ultimately degraded (Bergeron et al., 1985; Gordon et al., 1980). Some of the membrane-bound insulin is also degraded without requiring rapid internalization (Hamel et al., 1987).

Although the degradative pathway is not as well characterized, many lines of evidence now suggest that a metallo thiol

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<sup>\*</sup> Address correspondence to this author at the Section of Diabetes, Endocrinology, and Metabolism, University of Nebraska Medical Center, 42nd and Dewey Ave., Omaha, NE 68105.

<sup>&</sup>lt;sup>‡</sup>Veterans Administration Medical Center, Omaha, NE, and University of Nebraska Medical Center.

Present address: Fred Hutchinson Cancer Research Center, Seattle, WA 98104.

<sup>&</sup>lt;sup>1</sup> Veterans Administration Medical Center, Indianapolis, IN.

<sup>#</sup> Purdue University.

<sup>°</sup>Eli Lilly Research Laboratories.

<sup>&</sup>lt;sup>⋄</sup> Present address: The Ben May Institute, University of Chicago, Chicago, IL 60637.